Julius-Maximilians-UNIVERSITÄT WÜRZBURG

The Photophysics of Acenaphthylene

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Abstract

Acenaphthylene is a member of the group of polycyclic aromatic hydrocarbons (PAHs), which are formed in incomplete combustion processes but are also present in interstellar nurseries and play an important role as building blocks for modern organic devices. In ps time-resolved photoelectron imaging experiments we studied the photophysical behavior after excitation of the S₂ state. This short-lived electronic state first undergoes internal conversion (IC) to the S₁ state vibronic manifold, followed by another IC to the electronic ground state (S₀). This last step competes with intersystem crossing (ISC) to the triplet manifold, for which an energetic barrier has to be overcome to be enabled.



> [1+2'] and [1+2] REMPI

> Origins at 21757 (S₁) and 30076 cm⁻¹ (S₂)

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PHYSIKALISCHE UND

• C₁₂H₈

• M = 152 g/mol

• mp = $92^{\circ}C$

IE = 8.22 eV

THEORETISCHE CHEMIE

- Resolution of vibronic bands
- > S₂ spectrum is broadened
- > Totally symmetric vibrational modes

Photoelectron Imaging



Time-resolved PE-Imaging





- Pump probe spectroscopy
- > Transient photoelectron map
- Ionization via Rydberg states A and B
- > Time-dependent behavior
- > A and B vanish at long delay times
- > Unstructured transient signal remains
- \rightarrow No ionization from S₁/S₂
- \rightarrow Population of T-state (\rightarrow ISC)

Relaxation Scheme



Decay Traces



<u>S₁ state</u>

- > Mono exponential decay
- > Decreasing τ at higher excitation
- Signal drops to zero



- Additional fast time constant (< 1 ps)</p>
- > Further decrease of 2nd time constant
- Constant signal offset
- Long-lived component

- > Fast internal conversion $S_1 \leftarrow S_0 (IC_1)$
- \succ Competition IC₂ S₀ \leftarrow S₁ vs. ISC T_n \leftarrow S₁
- \succ No ISC after direct S₁ excitation
- > Energetic barrier has to be overcome

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Bunsentagung



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